







# Species diffusion coupled to elasticity in lithium ion batteries

Amabile Tatone   Chiara Mastrodicasa

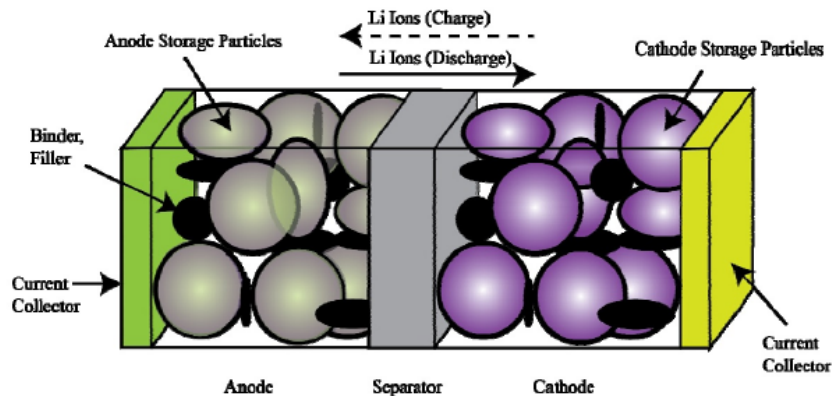
Department of Information Engineering,  
Computer Science and Mathematics,  
University of L'Aquila, Italy

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Lithium ion batteries

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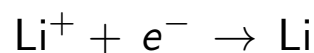
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[from: McMeeking-Purkayastha, Procedia IUTAM (2014)]

## Lithium ion battery

The main chemical reaction is the so-called *redox* reaction, which consists of a *reduction*



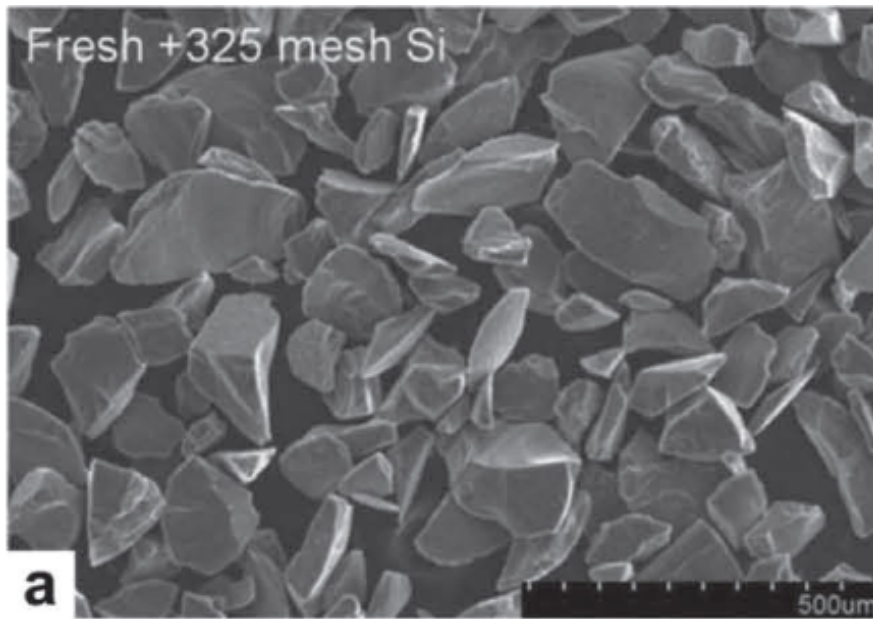
and an *oxidation*



In the charging process there is a *reduction* in the *anode* followed by *intercalation*.

The *intercalation* process makes the host crystal lattice swell, with the molar volume increasing several times.

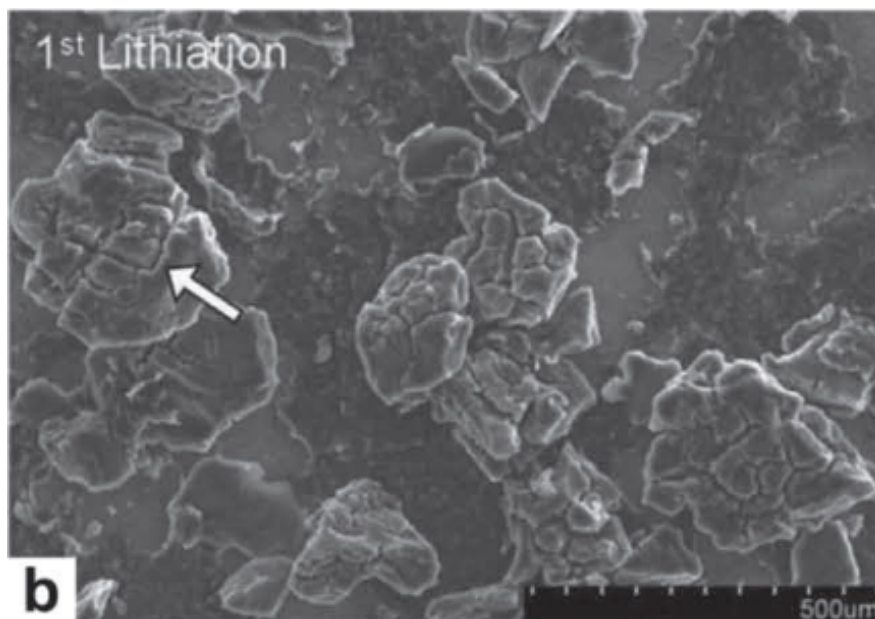
# Anode Si particles



[from: McDowell-Lee-Nix-Cui, Adv.Mater. (2013)]

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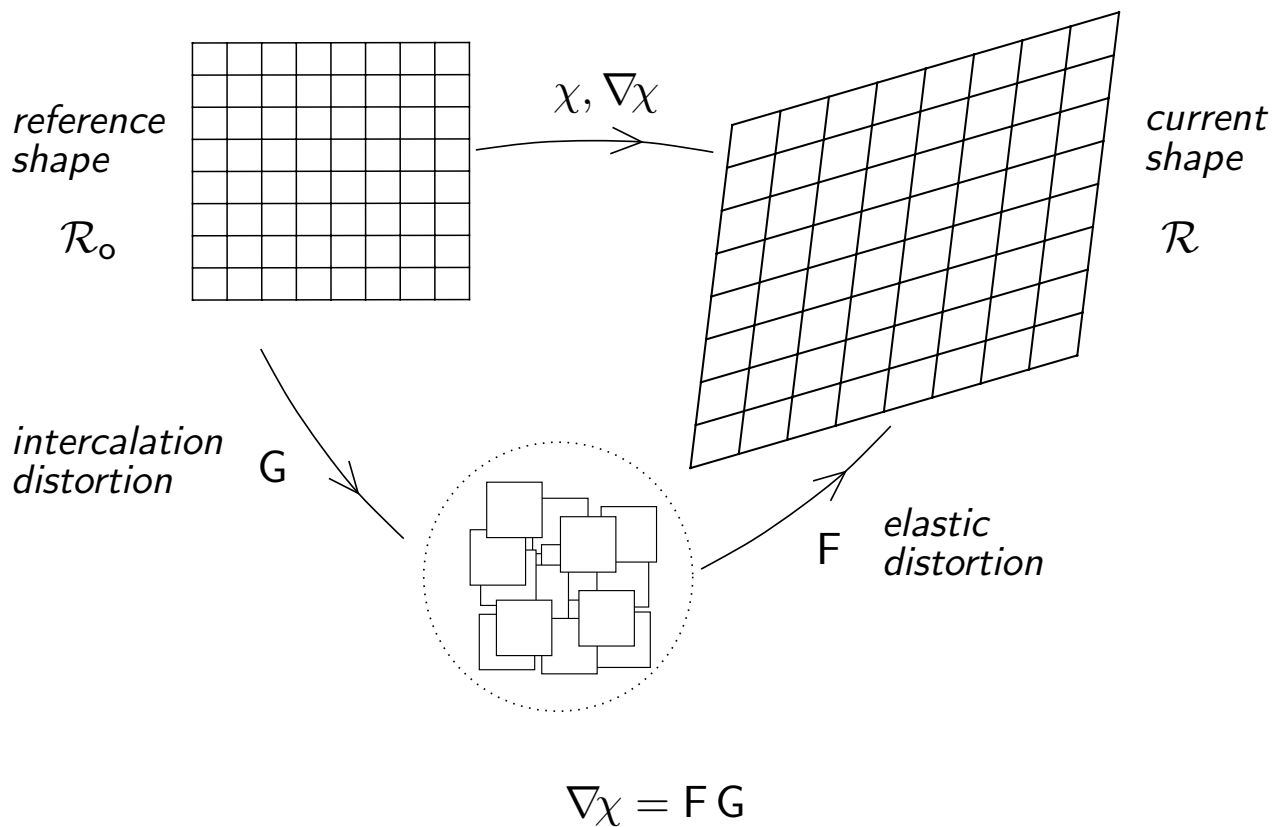
# Anode Si particles



[from: McDowell-Lee-Nix-Cui, Adv.Mater. (2013)]

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# Decomposition of the deformation gradient $\nabla\chi$



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## Kinematics and kinetics

Host material lattice deformation

$$\chi : \mathcal{R}_0 \rightarrow \mathcal{R}$$

$$\nabla\chi = FG$$

Lithium concentration

$$c = \frac{\rho_{Li}}{\rho_0}$$

$$G = \beta^{\frac{1}{3}} I$$

$$\beta = 1 + \alpha c$$

$$\frac{\rho_{Li}}{\rho_0} = \frac{\text{molar density of Li atoms per unit reference volume}}{\text{molar density of lattice sites per unit reference volume}}$$

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$$\frac{d}{dt} \int_{\mathcal{P}} \rho dV = \frac{d}{dt} \int_{\mathcal{P}_o} \rho_o dV = 0$$

$$\forall (\mathcal{P} \subset \mathcal{R}, \mathcal{P}_o \subset \mathcal{R}_o) \quad | \quad \chi: \mathcal{P}_o \rightarrow \mathcal{P}$$

$\rho$  molar density of lattice sites per unit current volume

$$\Rightarrow \dot{\rho} + \rho \operatorname{div} v = 0$$

$$\frac{d}{dt} \int_{\mathcal{P}} c \rho dV = \frac{d}{dt} \int_{\mathcal{P}_o} c \rho_o dV = \int_{\mathcal{P}_o} \dot{c} \rho_o dV = \int_{\mathcal{P}} \dot{c} \rho dV$$

$c \rho$  molar density of Li atoms per unit current volume

## Species molar balance

$$\frac{d}{dt} \int_{\mathcal{P}} c \rho dV = - \int_{\partial \mathcal{P}} h \cdot n dA + \int_{\mathcal{P}} h dV$$

$$\int_{\mathcal{P}} \dot{c} \rho dV = - \int_{\partial \mathcal{P}} h \cdot n dA + \int_{\mathcal{P}} h dV$$

$$\dot{c} \rho = - \operatorname{div} h + h$$

$$h = 0 \quad \Rightarrow \quad \dot{c} \rho = - \operatorname{div} h$$

$$\dot{c} \rho = - \operatorname{div} h$$

Introducing a scalar field  $\mu$  (energy per mole of lithium)  
transforming the *molar balance* into a *power balance*

$$\int_{\mathcal{P}} \mu \dot{c} \rho dV = - \int_{\mathcal{P}} \mu \operatorname{div} h dV \quad \forall \mu$$

$$\operatorname{div}(\mu h) = \mu \operatorname{div} h + \nabla \mu \cdot h$$

$$\int_{\mathcal{P}} \mu \dot{c} \rho dV = - \int_{\partial \mathcal{P}} \mu h \cdot n dA + \int_{\mathcal{P}} h \cdot \nabla \mu dV \quad \forall \mu$$

(  $\mu$  is the chemical potential )

## Species power balance

$$\int_{\mathcal{P}} \mu \dot{c} \rho dV = - \int_{\partial \mathcal{P}} \mu h \cdot n dA + \int_{\mathcal{P}} h \cdot \nabla \mu dV \quad \forall \mu$$

$$\int_{\mathcal{P}_o} \mu_o \dot{c} \rho_o dV = - \int_{\partial \mathcal{P}_o} \mu_o h_o \cdot n_o dA + \int_{\mathcal{P}_o} h_o \cdot \nabla \mu_o dV \quad \forall \mu_o$$

$$h_o = (\det F_o) F_o^{-1} h$$

$$\mu_o(x) = \mu(\chi(x))$$

Species power balance

$$\int_{\mathcal{P}_o} \mu_o \dot{c} \rho_o dV = - \int_{\partial\mathcal{P}_o} \mu_o \mathbf{h}_o \cdot \mathbf{n}_o dA + \int_{\mathcal{P}_o} \mathbf{h}_o \cdot \nabla \mu_o dV \quad \forall \mu_o$$

Force power balance

$$\int_{\mathcal{P}_o} \mathbf{b}_o \cdot \mathbf{v}_o dV + \int_{\partial\mathcal{P}_o} \mathbf{t}_o \cdot \mathbf{v}_o dA = \int_{\mathcal{P}_o} \mathbf{S}_o \cdot \nabla \mathbf{v}_o dV \quad \forall \mathbf{v}_o$$

## Free energy imbalance

Species power balance

$$\int_{\mathcal{P}_o} \mu_o \dot{c} \rho_o dV = - \underbrace{\int_{\partial\mathcal{P}_o} \mu_o \mathbf{h}_o \cdot \mathbf{n}_o dA}_{\text{external power}} + \int_{\mathcal{P}_o} \mathbf{h}_o \cdot \nabla \mu_o dV$$

Force power balance

$$\underbrace{\int_{\mathcal{P}_o} \mathbf{b}_o \cdot \mathbf{v}_o dV + \int_{\partial\mathcal{P}_o} \mathbf{t}_o \cdot \mathbf{v}_o dA}_{\text{external power}} = \int_{\mathcal{P}_o} \mathbf{S}_o \cdot \dot{\mathbf{F}}_o dV$$

Free energy imbalance

$$\mathbf{S}_o \cdot \dot{\mathbf{F}}_o + \mu_o \dot{c} \rho_o - \mathbf{h}_o \cdot \nabla \mu_o - \frac{d}{dt} \psi \geq 0$$

$$S_o \cdot \dot{F}_o + \mu_o \dot{c} \rho_o - h_o \cdot \nabla \mu_o - \frac{d}{dt} \psi \geq 0$$

$$\psi = \hat{\psi}(F_o, c)$$

$$\frac{d}{dt} \psi = S_o \cdot \dot{F}_o + \mu_o \dot{c} \rho_o$$

$$-h_o \cdot \nabla \mu_o \geq 0$$

$$h_o = -M \nabla \mu_o \quad (\text{Fick's law})$$

$$\hat{\psi}(F_o, c) = \rho_o \varphi_c(c) + \det G \varphi_e(F, c)$$

$$\hat{\psi}(F_o, c) = \det G \varphi(F)$$

$$\frac{d}{dt} \varphi(F) = S \cdot \dot{F}$$

$$F = F_o G^{-1} \quad G = \beta^{\frac{1}{3}} I \quad \beta = 1 + \alpha c$$

$$\frac{d}{dt} \psi = \beta S \cdot \dot{F} + \alpha \varphi(F) \dot{c}$$



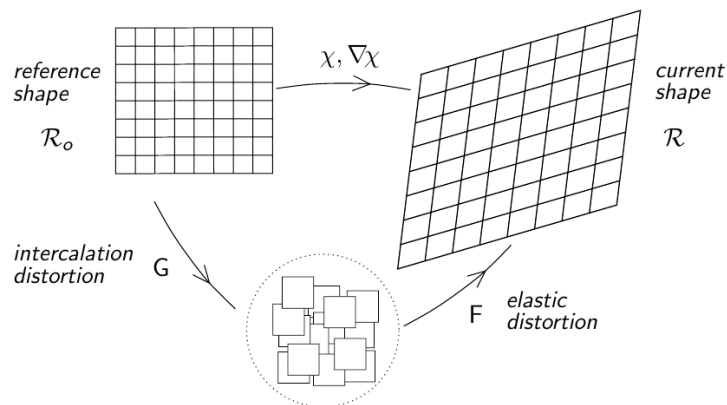
$$S_o \cdot \dot{F}_o + \mu_o \dot{c} \rho_o - h_o \cdot \nabla \mu_o - \frac{d}{dt} \psi \geq 0$$

$$\frac{d}{dt} \psi = \beta S \cdot \dot{F} + \alpha \varphi(F) \dot{c}$$

$$\frac{d}{dt} \psi = S_o \cdot \dot{F}_o + \mu_o \rho_o \dot{c} = \beta S \cdot \dot{F} + \frac{1}{3} \alpha S \cdot F \dot{c} + \mu_o \rho_o \dot{c}$$

$$-\alpha \varphi(F) + \frac{1}{3} \alpha \det F \operatorname{tr} T + \mu_o \rho_o = 0$$

# Constitutive characterization



$$\mu_o = \frac{\alpha}{\rho_o} \left( \varphi(F) - \frac{1}{3} \det F \operatorname{tr} T \right)$$

$$h_o = -M \nabla \mu_o$$

$$S \cdot \dot{F} = \frac{d}{dt} \varphi(F)$$

$$\varphi(\mathbf{F}) = k_I (\bar{\iota}_1 - 3) + k_V (J - 1)^2$$

$$\iota_1 = \text{tr } \mathbf{F}^T \mathbf{F}$$

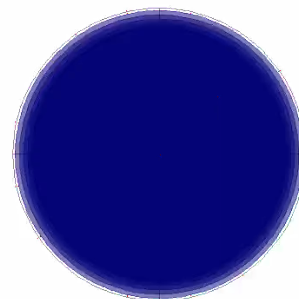
$$\bar{\iota}_1 = \iota_1 J^{-\frac{2}{3}}$$

$$J = \det \mathbf{F}$$

$$\hat{\mathbf{S}}(\mathbf{F}) = 2 k_I J^{-\frac{2}{3}} \left( \mathbf{F} - \frac{1}{3} \iota_1 \mathbf{F}^{-T} \right) + 2 k_V (J - 1) J \mathbf{F}^{-T}$$

$$\hat{\mathbf{T}}(\mathbf{F}) = 2 k_I J^{-\frac{5}{3}} \left( \mathbf{B} - \frac{1}{3} \iota_1 \mathbf{I} \right) + 2 k_V (J - 1) \mathbf{I}$$

## Constrained cylindrical particle



# Constrained cylindrical particle

Let us consider deformations such that

$$\mathbf{G} = \begin{pmatrix} \beta^{1/3} & 0 & 0 \\ 0 & \beta^{1/3} & 0 \\ 0 & 0 & \beta^{1/3} \end{pmatrix}$$

$$\mathbf{F}_o = \begin{pmatrix} \lambda_r & 0 & 0 \\ 0 & \lambda_r & 0 \\ 0 & 0 & \lambda \end{pmatrix}$$

$\lambda_r$  radial stretch

$\lambda$  axial stretch

with the constraint

$$\lambda_r = 1$$

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# Constrained cylindrical particle

As a consequence

$$\mathbf{F} = \begin{pmatrix} \beta^{-1/3} & 0 & 0 \\ 0 & \beta^{-1/3} & 0 \\ 0 & 0 & \lambda \beta^{-1/3} \end{pmatrix}$$

$$\varphi(\mathbf{F}) = k_I \left( \frac{2 + \lambda^2}{\lambda^{2/3}} - 3 \right) + k_V \left( \frac{\lambda}{\beta} - 1 \right)^2$$

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$$\hat{\sigma}_a(\lambda, \beta) = \frac{4}{3} k_I \beta \frac{\lambda^2 - 1}{\lambda^{5/3}} + 2 k_V \left( \frac{\lambda}{\beta} - 1 \right)$$

$$\hat{\sigma}_r(\lambda, \beta) = -\frac{2}{3} k_I \beta \frac{\lambda^2 - 1}{\lambda^{5/3}} + 2 k_V \left( \frac{\lambda}{\beta} - 1 \right)$$

$$\hat{\mu}_o(\lambda, \beta) = \frac{\alpha}{\rho_o} \left( k_I \left( \frac{\lambda^2 + 2}{\lambda^{2/3}} - 3 \right) - k_V \left( \frac{\lambda}{\beta} - 1 \right) \right)$$

$$\hat{\sigma}_a(\lambda, \beta) = 0 \quad \Rightarrow \quad \lambda = \hat{\lambda}(\beta)$$

$$\operatorname{div} \mathbf{h}_o = -\dot{c} \rho_o$$

$$\mathbf{h}_o = -M \nabla \mu_o$$

$$\mu_o = \hat{\mu}_o(\lambda, \beta)$$

# Constrained cylindrical particle

$$\hat{\sigma}_a(\lambda, \beta) = 0 \quad \Rightarrow \quad \lambda = \hat{\lambda}(\beta)$$

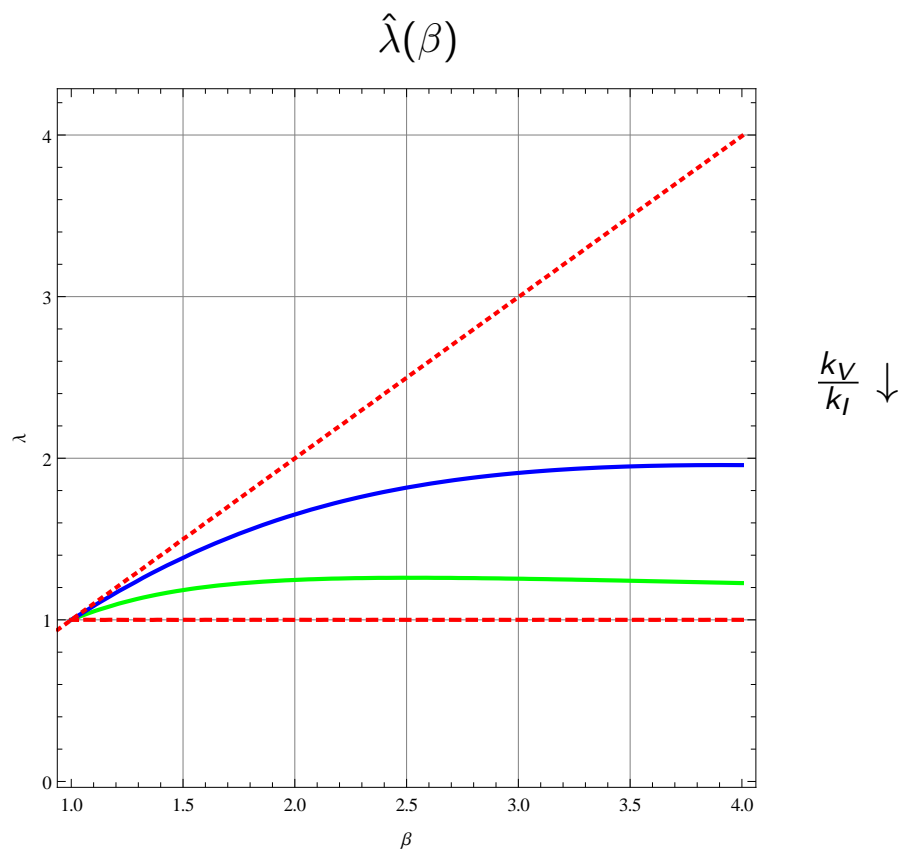
$$h'_o = -\dot{c} \rho_o$$

$$h_o = -d \mu'_o$$

$$\mu_o = \hat{\mu}_o(\lambda, \beta)$$

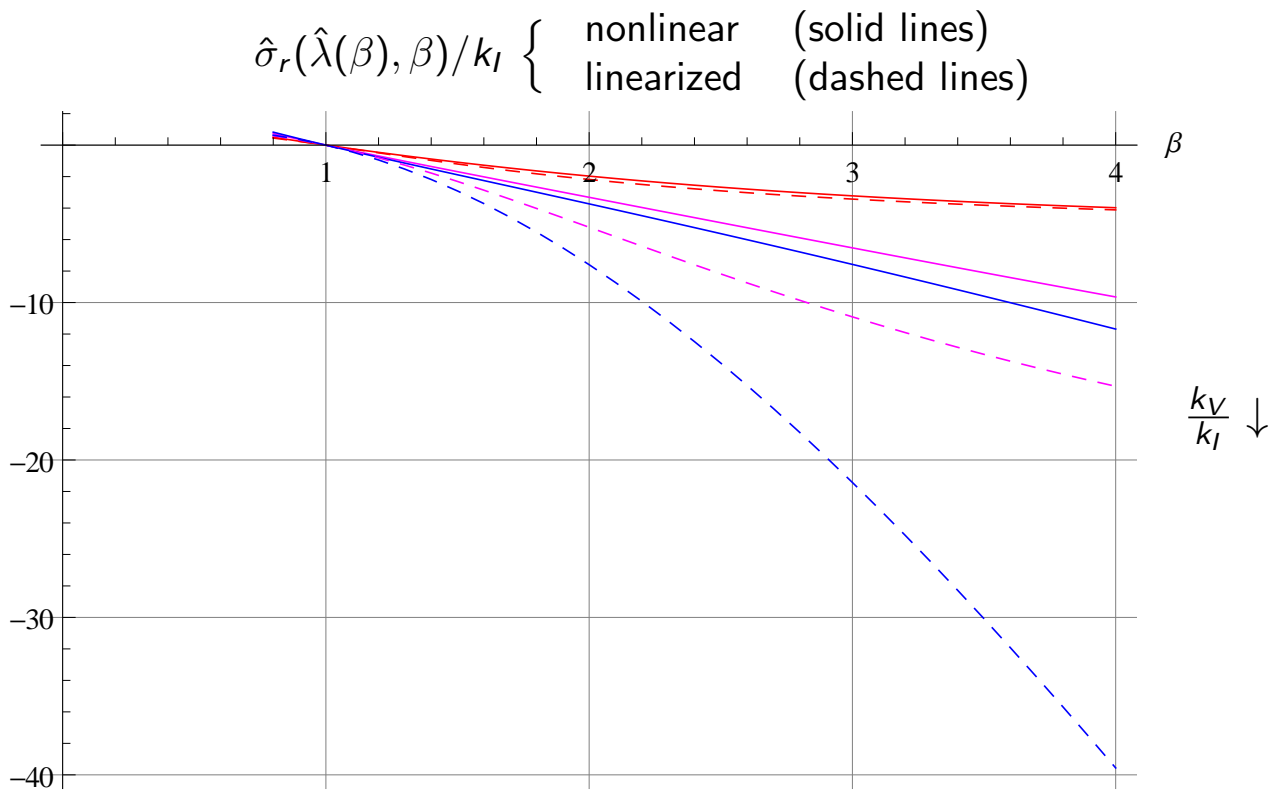
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# Constrained cylindrical particle



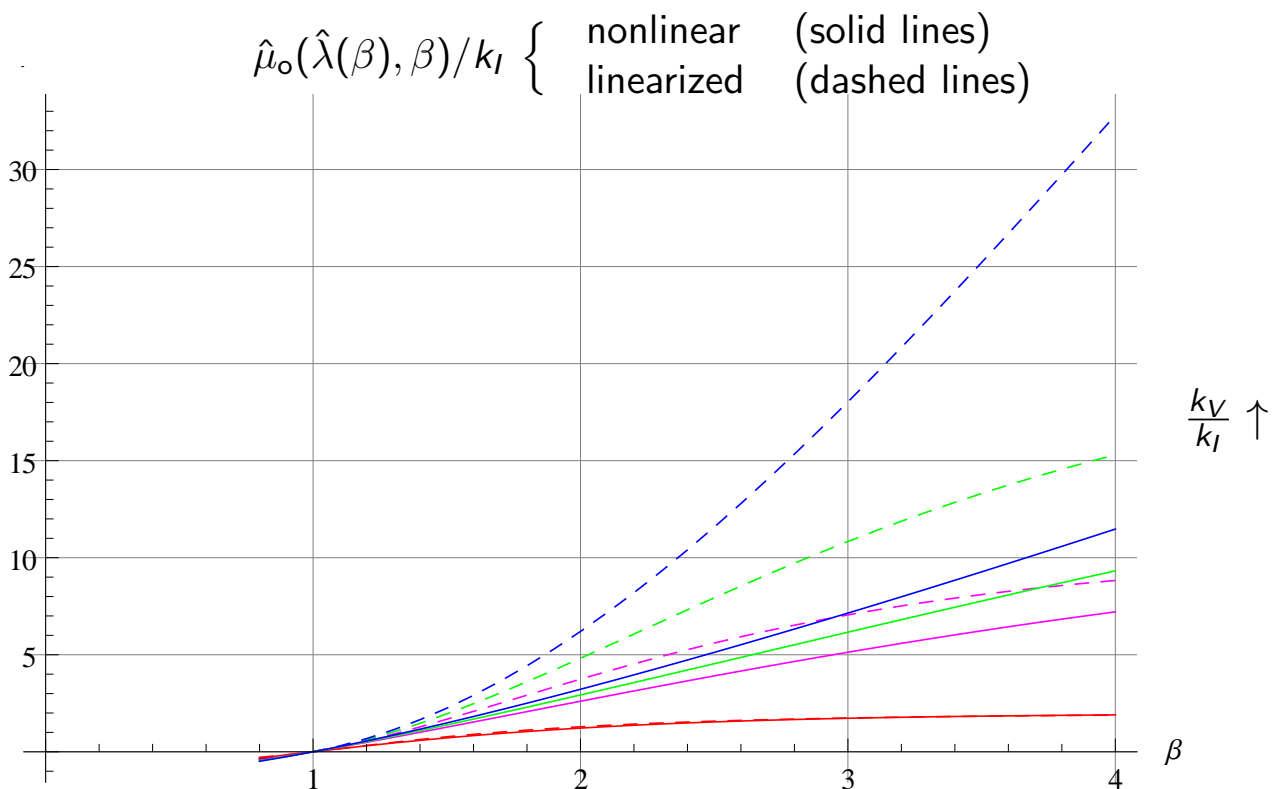
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# Constrained cylindrical particle



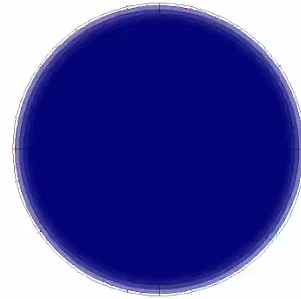
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# Constrained cylindrical particle



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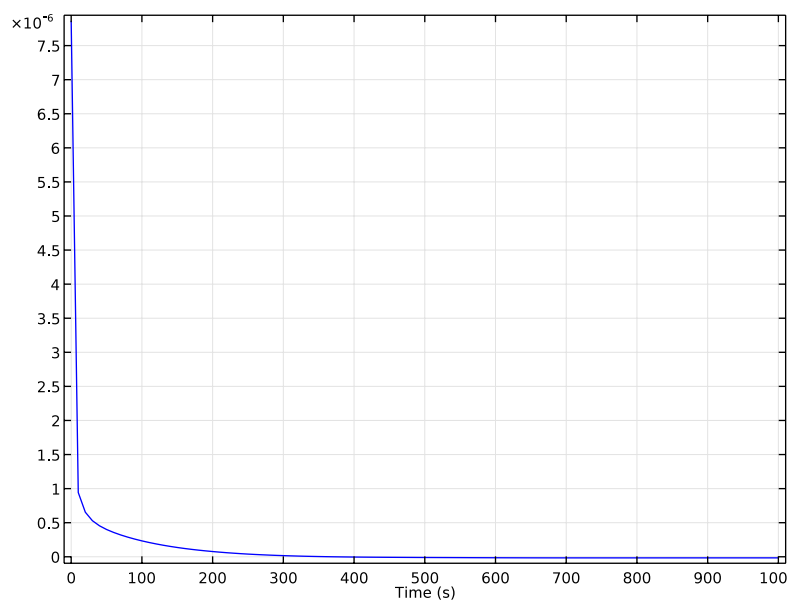
# Particle surrounded by a constant lithium concentration



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## *Cylindrical particle surrounded by a constant lithium concentration*

Total Li flux on the left face

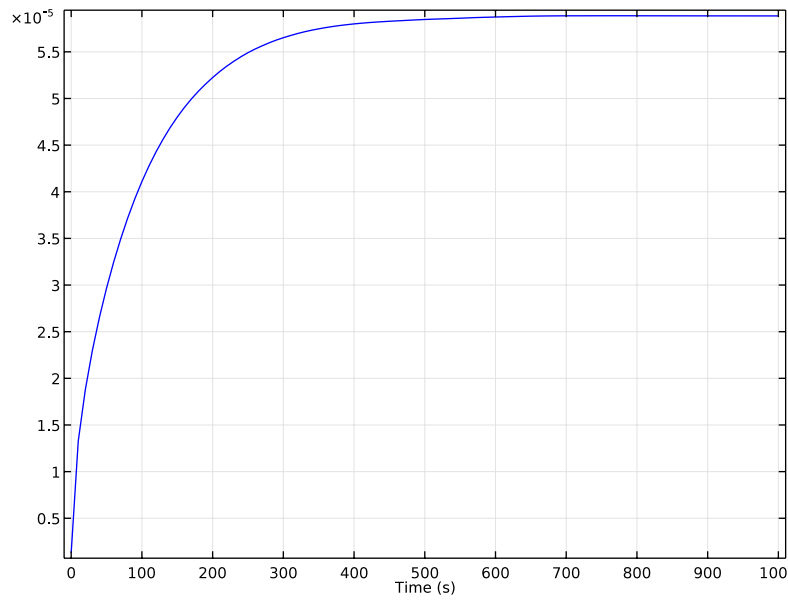


Comsol-constr-3D-cyl/batt-02b-cyl-m-pot



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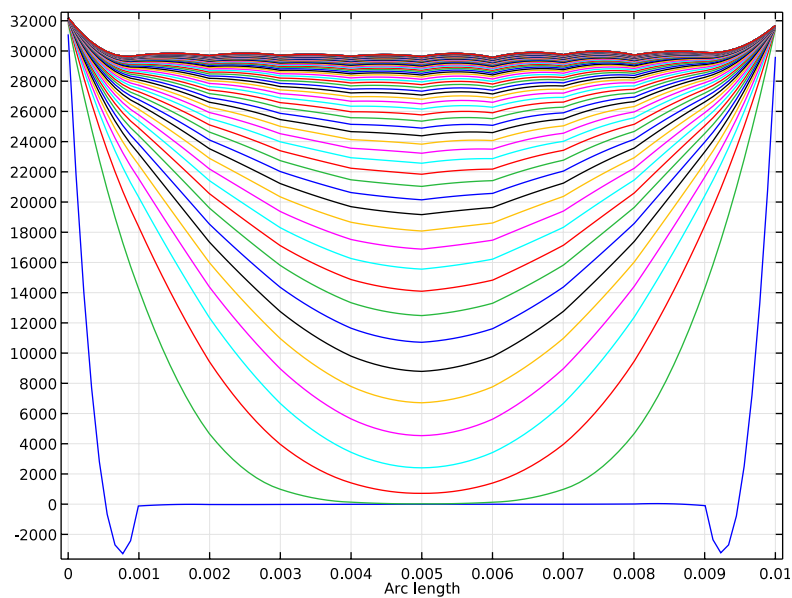
### Stored Li



Comsol-constr-3D-cyl/batt-02b-cyl-m-pot



### Chemical potential



Comsol-constr-3D-cyl/batt-02b-cyl-m-pot

